



# UNITED STATES PATENT AND TRADEMARK OFFICE

UNITED STATES DEPARTMENT OF COMMERCE  
United States Patent and Trademark Office  
Address: COMMISSIONER FOR PATENTS  
P.O. Box 1450  
Alexandria, Virginia 22313-1450  
www.uspto.gov

APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.
-----------------	-------------	----------------------	---------------------	------------------

09/810,962

03/16/2001

Atsuo Omaru

09792909-4809

7248

26263

7590

02/06/2009

SONNENSCHN NATH & ROSENTHAL LLP

P.O. BOX 061080

WACKER DRIVE STATION, SEARS TOWER

CHICAGO, IL 60606-1080

EXAMINER

DOVE, TRACY MAE

ART UNIT

PAPER NUMBER

1795

MAIL DATE

DELIVERY MODE

02/06/2009

PAPER

**Please find below and/or attached an Office communication concerning this application or proceeding.**

The time period for reply, if any, is set in the attached communication.

<b>Office Action Summary</b>	<b>Application No.</b> 09/810,962	<b>Applicant(s)</b> OMARU ET AL.	
	<b>Examiner</b> TRACY DOVE	<b>Art Unit</b> 1795	

**-- The MAILING DATE of this communication appears on the cover sheet with the correspondence address --**

**Period for Reply**

A SHORTENED STATUTORY PERIOD FOR REPLY IS SET TO EXPIRE 3 MONTH(S) OR THIRTY (30) DAYS, WHICHEVER IS LONGER, FROM THE MAILING DATE OF THIS COMMUNICATION.

- Extensions of time may be available under the provisions of 37 CFR 1.136(a). In no event, however, may a reply be timely filed after SIX (6) MONTHS from the mailing date of this communication.
- If NO period for reply is specified above, the maximum statutory period will apply and will expire SIX (6) MONTHS from the mailing date of this communication.
- Failure to reply within the set or extended period for reply will, by statute, cause the application to become ABANDONED (35 U.S.C. § 133). Any reply received by the Office later than three months after the mailing date of this communication, even if timely filed, may reduce any earned patent term adjustment. See 37 CFR 1.704(b).

**Status**

- 1) ☒ Responsive to communication(s) filed on 18 November 2008.
- 2a) ☒ This action is **FINAL**.                      2b) ☐ This action is non-final.
- 3) ☐ Since this application is in condition for allowance except for formal matters, prosecution as to the merits is closed in accordance with the practice under *Ex parte Quayle*, 1935 C.D. 11, 453 O.G. 213.

**Disposition of Claims**

- 4) ☒ Claim(s) 1,3-5 and 15-46 is/are pending in the application.
- 4a) Of the above claim(s) 15-46 is/are withdrawn from consideration.
- 5) ☐ Claim(s) \_\_\_\_\_ is/are allowed.
- 6) ☒ Claim(s) 1 and 3-5 is/are rejected.
- 7) ☐ Claim(s) \_\_\_\_\_ is/are objected to.
- 8) ☐ Claim(s) \_\_\_\_\_ are subject to restriction and/or election requirement.

**Application Papers**

- 9) ☐ The specification is objected to by the Examiner.
- 10) ☐ The drawing(s) filed on \_\_\_\_\_ is/are: a) ☐ accepted or b) ☐ objected to by the Examiner.  
Applicant may not request that any objection to the drawing(s) be held in abeyance. See 37 CFR 1.85(a).  
Replacement drawing sheet(s) including the correction is required if the drawing(s) is objected to. See 37 CFR 1.121(d).
- 11) ☐ The oath or declaration is objected to by the Examiner. Note the attached Office Action or form PTO-152.

**Priority under 35 U.S.C. § 119**

- 12) ☐ Acknowledgment is made of a claim for foreign priority under 35 U.S.C. § 119(a)-(d) or (f).
- a) ☐ All    b) ☐ Some \*    c) ☐ None of:
1. ☐ Certified copies of the priority documents have been received.
2. ☐ Certified copies of the priority documents have been received in Application No. \_\_\_\_\_.
3. ☐ Copies of the certified copies of the priority documents have been received in this National Stage application from the International Bureau (PCT Rule 17.2(a)).

\* See the attached detailed Office action for a list of the certified copies not received.

**Attachment(s)**

- |  |   |
|--|---|
| 1) <input type="checkbox"/> Notice of References Cited (PTO-892)                     | 4) <input type="checkbox"/> Interview Summary (PTO-413)           |
| 2) <input type="checkbox"/> Notice of Draftsperson's Patent Drawing Review (PTO-948) | Paper No(s)/Mail Date. _____                                      |
| 3) <input type="checkbox"/> Information Disclosure Statement(s) (PTO/SB/08)          | 5) <input type="checkbox"/> Notice of Informal Patent Application |
| Paper No(s)/Mail Date _____  | 6) <input type="checkbox"/> Other: _____                          |

### **DETAILED ACTION**

This Office Action is in response to the communication filed on 11/18/08. Applicant's arguments have been considered, but are not persuasive. Claims 1, 3-5 and 15-46 are pending with claims 15-46 being withdrawn. This Action is FINAL.

#### ***Claims Analysis***

The claimed invention recites the graphite in the negative electrode has a "rhombohedral structure". The specification discloses that natural graphite has a "rhombohedral structure" (page 21) and that natural graphite having a "rhombohedral structure" may be used as a starting material (page 42). Thus, in view of the teaching of the present specification, natural graphite contains a "rhombohedral structure".

#### ***Claim Rejections - 35 USC § 112***

The following is a quotation of the first paragraph of 35 U.S.C. 112:

The specification shall contain a written description of the invention, and of the manner and process of making and using it, in such full, clear, concise, and exact terms as to enable any person skilled in the art to which it pertains, or with which it is most nearly connected, to make and use the same and shall set forth the best mode contemplated by the inventor of carrying out his invention.

Claim 1 is rejected under 35 U.S.C. 112, first paragraph, because the specification, while being enabling for a graphite material having the claimed relationship for Gs, does not reasonably provide enablement for a negative electrode having the claimed relationship Gs. The specification does not enable any person skilled in the art to which it pertains, or with which it is most nearly connected, to make or use the invention commensurate in scope with these claims. Note if the claims were amended to recite the negative electrode consists of graphite, this rejection would be withdrawn.

Claims 1 and 3-5 are rejected under 35 U.S.C. 112, first paragraph, as failing to comply with the written description requirement. The claim(s) contains subject matter which was not described in the specification in such a way as to reasonably convey to one skilled in the relevant art that the inventor(s), at the time the application was filed, had possession of the claimed invention. Each of claims 1 and 3-5 recite “the negative electrode is pressed such that the specific surface area is increased by *at least* 2.5 times that of the specific surface area before being pressed”, which does not appear to be supported by the specification as filed. See at least the abstract and [0010]. Furthermore, the specification discloses it is the graphite material that is “pressed”, not the negative electrode.

The following is a quotation of the second paragraph of 35 U.S.C. 112:

The specification shall conclude with one or more claims particularly pointing out and distinctly claiming the subject matter which the applicant regards as his invention.

Claims 1 and 3-5 are rejected under 35 U.S.C. 112, second paragraph, as being indefinite for failing to particularly point out and distinctly claim the subject matter which applicant regards as the invention.

Each of claims 1 and 3-5 recite “the negative electrode is pressed such that the specific surface area is increased by at least 2.5 times that of the specific surface area before being pressed”, which is indefinite. It is unclear what the specific surface area value range of the claimed material encompasses.

To the extent the claims are understood in view of the 35 USC 112 rejections above, note the following prior art rejections.

***Claim Rejections - 35 USC § 103***

Art Unit: 1795

The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:

(a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negated by the manner in which the invention was made.

Claims 1 and 3-5 are rejected under 35 U.S.C. 103(a) as being unpatentable over Hayashi et al., JP 10-334915.

Hayashi teaches a rechargeable battery having an electrode comprising graphite particles. A dynamic energy process is applied to a graphite material so that the apparent density ratio between before and after the process becomes 1.1 or above. The apparent density ratio between before and after the process equals the tap density after the process/tap density before the process, and this is to become the index of sphericity. See abstract.

The intensity ratio  $R$  of a Raman spectrum is preferably 0.4 or less. In the Raman spectrum analysis, the intensity  $I_A$  of peak  $PA$  near  $1580\text{ cm}^{-1}$  and the intensity  $I_B$  of peak  $PB$  near  $1360\text{ cm}^{-1}$  were measured (0035). Therefore,  $R = I_B/I_A = H_{sd}/H_{sg}$  and  $H_{sg}/H_{sd} = 1/R = G_s$ . Since  $R$  is 0.4 or less, Hayashi teaches  $G_s$  is 2.5 or more.

The tap density ratio before and after processing is 1.7 or greater, more preferably 1.1 or greater. It is desirable to have a tap density after processing of 0.5-2 g/cc (see page 4, paragraph 0023-0024). The tap density of the graphite material is preferably in the range of 0.7-1.2 g/cc (see page 7, paragraph 0042). The true density of the graphite material is 2.25 g/cc or more (claim 2). Thus a packing characteristic

Art Unit: 1795

index (tap density/true density) of Hayashi may be 0.53 ( $1.2/2.25$  = tap density/true density).

The specific surface area of the graphite particles after processing (pulverizing) is below  $25 \text{ m}^2/\text{g}$  and more than  $0.5 \text{ m}^2/\text{g}$ , preferably  $2\text{-}10 \text{ m}^2/\text{g}$  (0035). Table 4 shows different graphite material properties before and after a dynamic energy process/treatment. The SA in Table 4 represents surface area with the surface area of the graphite being  $19.1 \text{ m}^2/\text{g}$  before treatment and  $8.9 \text{ m}^2/\text{g}$  after treatment (Example 13). The surface area after treatment is 2.1 times that before treatment. The energy process is specifically pulverization. Hayashi teaches a surface area of the graphite being  $4.5 \text{ m}^2/\text{g}$ ,  $4.8 \text{ m}^2/\text{g}$ ,  $8.7 \text{ m}^2/\text{g}$  or  $19.1 \text{ m}^2/\text{g}$  before treatment (Table 4) and preferably  $2\text{-}10 \text{ m}^2/\text{g}$  after treatment (0035). Hayashi teaches an electrode having a graphite material with a (d002) distance between layers of  $0.34\text{nm}$  or less (claim 2).

Hayashi teaches natural graphite of high orientation/high crystallinity is used (0013-0014). High crystallinity natural graphite is known to have a rhombohedral structure (diamond structure). Hayashi teaches the natural graphite may be subjected to a surface grinding process (0029). Natural graphite has a rhombohedral structure (as stated in the present specification, see above). A polypropylene separator is disclosed at [0043].

Hayashi does not explicitly recite the graphite material has at least two peaks on a differential thermogravimetric curve. However, the graphite material of Hayashi inherently has at least two peaks on a differential thermogravimetric curve because the graphite material of Hayashi has a Raman spectrum having two distinct signal peaks.

Art Unit: 1795

The two distinct signal peaks on the Raman spectrum indicate the graphite material contains two distinct carbon materials. A graphite material having two distinct carbon materials would inherently provide at least two peaks on a differential thermogravimetric curve. Hayashi does not explicitly state the weight rejection as measured by DTG, is at least 5% and at most 40%. However, the limitation is a product-by-process limitation, which is not given patentable weight in the absence of unexpected results.

The limitation regarding stacking band shaped layers of the battery is well known in the art. A battery must have a positive electrode and a negative electrode that are separated. Stacking a band positive electrode, a first separator, a band negative electrode and a second separator was not invented by Applicant. This is a well known battery structure.

Claims 1 and 3-5 are rejected under 35 U.S.C. 103(a) as being unpatentable over Morita et al., EP 0861804 A1.

Morita teaches a nonaqueous electrolyte secondary battery having a positive electrode, a negative electrode and a polypropylene separator (9:39-10:52). The negative electrode includes a carbon coated carbonaceous core material. The core material may be natural graphite (rhombohedral structure) having a ratio of peak strength around  $1360\text{ cm}^{-1}$  to peak strength around  $1580\text{ cm}^{-1}$  determined by Raman spectroscopy of 0.5 or less (4:25-46). The carbon material has a specific surface area of  $5\text{ m}^2/\text{g}$  or less, preferably about  $1\text{-}5\text{ m}^2/\text{g}$  (3:8-13). The polypropylene separator is between the positive and negative electrodes (8:4-13). The coated carbon material has a true density of about  $1.5\text{-}2.26\text{ g/cm}^3$ , preferably about  $1.8\text{-}2.26\text{ g/cm}^3$  (5:21-24).

Art Unit: 1795

Morita does not explicitly recite the graphite material has at least two peaks on a differential thermogravimetric curve. However, the graphite material of Morita inherently has at least two peaks on a differential thermogravimetric curve because the graphite material of Morita has a Raman spectrum having two distinct signal peaks. The two distinct signal peaks on the Raman spectrum indicate the graphite material contains two distinct carbon materials. A graphite material having two distinct carbon materials would inherently provide at least two peaks on a differential thermogravimetric curve. Morita does not explicitly state the weight rejection as measured by DTG, is at least 5% and at most 40%. However, the limitation is a product-by-process limitation, which is not given patentable weight in the absence of unexpected results.

The limitation regarding stacking band shaped layers of the battery is well known in the art. A battery must have a positive electrode and a negative electrode that are separated. Stacking a band positive electrode, a first separator, a band negative electrode and a second separator was not invented by Applicant. This is a well known battery structure.

### ***Response to Arguments***

Applicant's arguments filed 11/18/08 have been fully considered but they are not persuasive. Applicant argues Hayashi fails to disclose or suggest a band shaped negative electrode pressed such that the specific surface area of the negative electrode is increased by at least 2.5 times that of the specific surface area before being pressed. Note this limitation has been rejected under 35 USC 112. Furthermore, Hayashi teaches the specific surface area of the graphite particles after processing (pulverizing)



Art Unit: 1795

is below  $25 \text{ m}^2/\text{g}$  and more than  $0.5 \text{ m}^2/\text{g}$ , preferably  $2\text{-}10 \text{ m}^2/\text{g}$  (0035). Table 4 shows different graphite material properties before and after a dynamic energy process/treatment. The SA in Table 4 represents surface area with the surface area of the graphite being  $19.1 \text{ m}^2/\text{g}$  before treatment and  $8.9 \text{ m}^2/\text{g}$  after treatment (Example 13). The surface area after treatment is 2.1 times that before treatment. The energy process is specifically pulverization. Hayashi teaches a surface area of the graphite being  $4.5 \text{ m}^2/\text{g}$ ,  $4.8 \text{ m}^2/\text{g}$ ,  $8.7 \text{ m}^2/\text{g}$  or  $19.1 \text{ m}^2/\text{g}$  before treatment (Table 4) and preferably  $2\text{-}10 \text{ m}^2/\text{g}$  after treatment (0035).

Applicant argues Hayashi does not teach the stacking order of the claimed invention. However, the limitation regarding stacking band shaped layers of the battery is well known in the art. A battery must have a positive electrode and a negative electrode that are separated. Stacking a band positive electrode, a first separator, a band negative electrode and a second separator was not invented by Applicant. This is a well known battery structure.

### ***Conclusion***

Applicant's amendment necessitated the new ground(s) of rejection presented in this Office action. Accordingly, **THIS ACTION IS MADE FINAL**. See MPEP § 706.07(a). Applicant is reminded of the extension of time policy as set forth in 37 CFR 1.136(a).

A shortened statutory period for reply to this final action is set to expire **THREE MONTHS** from the mailing date of this action. In the event a first reply is filed within **TWO MONTHS** of the mailing date of this final action and the advisory action is not

Art Unit: 1795

mailed until after the end of the THREE-MONTH shortened statutory period, then the shortened statutory period will expire on the date the advisory action is mailed, and any extension fee pursuant to 37 CFR 1.136(a) will be calculated from the mailing date of the advisory action. In no event, however, will the statutory period for reply expire later than SIX MONTHS from the date of this final action.

Any inquiry concerning this communication or earlier communications from the examiner should be directed to Tracy Dove whose telephone number is 571-272-1285. The examiner can normally be reached on Monday-Thursday (9:00-7:30).

If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Pat Ryan can be reached on 571-272-1292. The fax phone number for the organization where this application or proceeding is assigned is 703-872-9306.

Information regarding the status of an application may be obtained from the Patent Application Information Retrieval (PAIR) system. Status information for published applications may be obtained from either Private PAIR or Public PAIR. Status information for unpublished applications is available through Private PAIR only. For more information about the PAIR system, see <http://pair-direct.uspto.gov>. Should you have questions on access to the Private PAIR system, contact the Electronic Business Center (EBC) at 866-217-9197 (toll-free).

January 29, 2009

/TRACY DOVE/

Primary Examiner, Art Unit 1795